

Surfactant-Induced Modulation of Fluorosensor Activity: A Simple Way to Maximize the Sensor Efficiency [*J. Am. Chem. Soc.* 2006, *128*, 3126–3127]. Arabinda Mallick, Malay C. Mandal, Basudeb Haldar, Alok Chakrabarty, Paramita Das, and Nitin Chattopadhyay*

While repeating and extending the work using sodium dodecyl sulfates (SDS) obtained from USB and Fluka, we found that the SDS that we used for our published article was not up to the mark, and we want to replace Figure 2 by the following figure. Although the pattern of the two figures is essentially the same, the optimum concentration of SDS for attaining the maximum sensing efficiency is shifted from 2.5 mM to 6.5 mM. Further, the reported enhancement factor in K_{SV} at the optimum SDS concentration should be rectified to a value of 1000, and the K_{SV} and k_q values will be modified accordingly.

The corrections, however, affect neither the concept of maximizing the sensor efficiency by the use of surfactants nor the impact of the published work.



Figure 2. Variation of K_{SV} as a function of SDS concentration.

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It has come to our attention that the chemical formula of the rhodium complex used in this report was inadvertently misrepresented, both in the main text (the third paragraph of page 6336, the equation of Table 1, and the footnote of Table 2) and in the Supporting Information. The chemical formula of the rhodium complex should be uniformly corrected to chlorobis(ethylene)-rhodium(I) dimer, $[Rh(C_2H_4)_2Cl]_2$, not chlorobis(acetylene)-rhodium(I) dimer, $[Rh(C_2H_2)_2Cl]_2$, as it was originally written. The corrected version of the Supporting Information is provided accordingly. We regret these mistakes.

Supporting Information Available: Experimental procedures and spectral data for all new compounds (corrected PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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